

# The influence of the liquid slab thickness on the planar vapor-liquid interfacial tension

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## Abstract

One of the long standing challenges in molecular simulation is the description of interfaces. On the molecular length scale, finite size effects significantly influence the properties of the interface such as its interfacial tension, which can be reliably investigated by molecular dynamics simulation of planar vapor-liquid interfaces. For the Lennard-Jones fluid, finite size effects are examined here by varying the thickness of the liquid slab. It is found that the surface tension and density in the center of the liquid region decreases significantly for thin slabs. The influence of the slab thickness on both the liquid density and the surface tension is found to scale with  $1/S^3$  in terms of the slab thickness  $S$ , and a linear correlation between both effects is obtained. The results corroborate the analysis of Malijevský and Jackson, *J. Phys.: Cond. Mat.* 24: 464121 (2012), who recently detected an analogous effect for the surface tension of liquid nanodroplets.

*Keywords:* Surface tension, long range correction, finite size effects

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## 1. Introduction

Molecular simulation is a well-established approach for the analysis of fluid interfaces and their molecular structure. Much work has been dedicated to the interfacial tension.<sup>1-4</sup> For a fluid interface, there are (at least) three different aspects in which its size can be varied, each of which may affect the interfacial tension:

- curvature effects, depending on the local characteristic radii of curvature
- capillary wave effects, depending on the range of wavelengths permitted by the morphology and size of the interface
- confinement effects, which arise due to spatial restrictions imposed on a fluid phase by one or several interfaces or walls

According to the Tolman<sup>5</sup> approach, the interfacial tension of a nanodroplet deviates from that of a planar interface due to its extremely curved shape.<sup>6-10</sup> However, it should be noted that all three phenomena are present when the size of a droplet is varied: Smaller droplets have a higher curvature, a smaller range of capillary wavelengths, and a more significant deviation from bulk-like behaviour due to confinement. In addition to curvature, the other effects might therefore also have a significant influence on the formation of droplets in a supersaturated vapor, where the size of the critical nucleus and the

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21 nucleation rate are of major interest.<sup>11–15</sup> A similar case is cavitation, where  
22 bubbles emerge in a liquid phase.<sup>16–18</sup>

23 Spherically curved interfaces of droplets were simulated for the first time  
24 by molecular simulation in the early 1970s.<sup>19,20</sup> Nonetheless, while curvature  
25 effects are relatively well-studied, there are only astonishingly few investi-  
26 gations related to the other size effects, which are also present in the case  
27 of planar interfaces. Several previous works address the influence of small  
28 simulation volumes,<sup>21–24</sup> which is usually discussed in terms of capillary wave  
29 effects.<sup>24–28</sup> The present study considers the influence of the liquid slab thick-  
30 ness, i.e. of confinement by two parallel planar vapor-liquid interfaces which  
31 are close to each other. This effect was previously investigated by Weng et  
32 al.<sup>29</sup> who, however, did not find a systematic correlation.

33 The computation of interfacial properties is always done in a single sim-  
34 ulation volume containing both phases, the liquid and the vapor phase, sep-  
35 arated by the interface. For the calculation of the bulk properties there are  
36 many other methods, like Grand Equilibrium,<sup>30</sup>  $NpT$  plus test particle,<sup>31</sup>  
37 or the Gibbs ensemble method.<sup>32</sup> The surface tension can be computed for  
38 example via the virial route or the surface free energy.<sup>2,33–37</sup> The virial route  
39 is directly related to the common approach for calculating the pressure in a  
40 molecular simulation.<sup>22,23,38,39</sup> It is known that the pressure in dependence  
41 of the density exhibits a van der Waals loop in the two phase region.<sup>40,41</sup>

42 At interfaces, the long range contribution to the interaction potential  
43 plays an important role for all thermodynamic properties.<sup>42–44</sup> Nonetheless,  
44 there are also simulations applying a truncated and shifted potential, which  
45 neglects the whole long range contribution.<sup>16,45–47</sup> When dealing with a ho-

46 homogeneous system, long range corrections are only needed for energy and  
47 pressure,<sup>48,49</sup> while in an inhomogeneous configuration, also the dynamics of  
48 the systems needs to be appropriately corrected.<sup>50-52</sup>

49 In the present work, the influence of the liquid slab thickness on thermo-  
50 dynamic properties is discussed. A suitable long range correction is used to  
51 obtain accurate and validated results.

## 52 2. Simulation method

53 In this study, the Lennard-Jones potential (LJ)

$$u_{ij} = 4\epsilon \left[ \left( \frac{\sigma}{r_{ij}} \right)^{12} - \left( \frac{\sigma}{r_{ij}} \right)^6 \right] \quad (1)$$

54 is employed, where  $\epsilon$  and  $\sigma$  are the energy and size parameters and  $r_{ij}$  is  
55 the distance between the two particles  $i$  and  $j$ . The standard LJ parameters  
56  $\epsilon = 1$  and  $\sigma = 1$  are used, yielding a reduced LJ potential.

57 As usual, the potential was truncated in order to reduce the computing  
58 time. A cutoff radius of  $r_c = 3$  was used for the present simulations. To  
59 diminish the error made by this assumption, a bin based tail correction was  
60 applied to the simulation.<sup>50,53</sup> Thereby, the potential energy, the forces acting  
61 on the molecules, and the virial are each split into an explicitly computed  
62 part and a long range correction. The calculation of the correction terms  
63 was conducted every 10 time steps. It is known that this method provides  
64 cutoff independent results for the LJ fluid.<sup>50</sup> For a discussion of the employed  
65 method in full detail, the reader is referred to Janeček's work,<sup>50</sup> wherein this  
66 approach was first presented.

67 The interfacial tension  $\gamma$  is given by the difference between the diagonal  
 68 components of the virial tensor  $\Pi_N - \Pi_T$  or, equivalently, an integral over  
 69 the differential pressure<sup>50,54</sup>  $p_N - p_T$

$$\gamma = \frac{1}{2A} (\Pi_N - \Pi_T) = \int_{-\infty}^{\infty} (p_N - p_T) dy, \quad (2)$$

70 where  $2A$  denotes the interfacial area of the two interfaces. The pressure  
 71 calculation is based on the method proposed by Irving and Kirkwood,<sup>55</sup> but  
 72 in contrast to their approach the pressure is not homogeneously distributed  
 73 between the positions of the two particles. To speed up the simulation the  
 74 pressure is divided between the bins of the involved particles. It should be  
 75 noted that this modification has a slight effect on the localized pressure tensor  
 76 but leads to the same outcome for the overall surface tension.<sup>23</sup>

77 Molecular dynamics (MD) simulations were conducted in the canonical  
 78 ensemble with  $N = 16\,000$  particles. The equimolar thickness of the li-  
 79 quid slab  $S$  was varied between 12 and the minimum stable configuration.  
 80 The equimolar thickness was determined using the saturated densities  $\rho'$  and  
 81  $\rho''$  for the given temperature, the simulation volume  $V$ , and the number of  
 82 particles  $N$

$$S = \frac{N - \rho''V}{(\rho' - \rho'')A}, \quad (3)$$

83 i.e.  $S$  only depends on the boundary conditions applied to the molecular sim-  
 84 ulation within the canonical ensemble, not on the outcome of the simulation,  
 85 and it does not vary over simulation time. The temperature was kept con-  
 86 stant by a velocity scaling thermostat. All simulations were performed in a  
 87 parallelepiped box. The elongation of the simulation volume in  $y$  direction,  
 88 i.e. normal to the interface, was  $\ell_y = 50$ .

89 For the simulation of a reference case the number of particles  $N$  was  
90 increased to 300 000, the elongation in  $y$  direction was  $\ell_y = 100$  and a slab  
91 thickness of  $S = 40$  was used. The temperature  $T$  was varied between 0.7 and  
92 1.25, i.e. from the triple point temperature to 95% of the critical temperature.  
93 The simulations were carried out using the *ls1* molecular dynamics code.<sup>56</sup>  
94 The equation of motion was solved by a leapfrog integrator.<sup>57</sup> A time step of  
95  $\Delta t = 0.002$  was used. The equilibration was conducted for at least 120 000  
96 time steps, while the production ran for 840 000 time steps. The statistical  
97 errors given in the present study are equal to 3 times the standard deviation  
98 of 7 block averages, each over 120 000 time steps.

99 A further series of simulations was conducted to validate whether rela-  
100 tively small cutoff radii are permissible when Janeček's cutoff correction is  
101 employed. The simulation results support this conclusion, cf. Fig. 1.

102 [Figure 1 about here.]

### 103 3. Results

104 [Figure 2 about here.]

105 [Figure 3 about here.]

106 As described above, a series of simulations was carried out using a large liquid  
107 slab of  $S = 40$  in order to approximate bulk phase behavior. The resulting  
108 values  $\gamma_\infty$  and  $\rho'_\infty$  are used as a reference for the further simulations. The  
109 resulting surface tension is shown in Fig. 2. The regression

$$\gamma = 2.94 \left(1 - \frac{T}{T_c}\right)^{1.23} \quad (4)$$

110 is obtained, with  $T_c = 1.3126$  according to Pérez Pellitero et al.<sup>58</sup> The type  
111 of correlation is the same as proposed by Vrabec et al.<sup>46</sup> for the truncated  
112 and shifted LJ potential, and their exponent (i.e. 1.21) is also very similar to  
113 the present one.

114 Moreover, simulations were also performed for smaller slab thicknesses  
115 ( $S \leq 12$ ). Thereby, MD runs were conducted with successively smaller values  
116 of  $S$ , until a minimum stable thickness was reached for the given temperature.

117 In Fig. 3, the density and the corresponding differential pressure profile  
118 for  $T = 0.7$  is plotted over the  $y$  coordinate. It can be seen that the density  
119 in the center of the slab for  $S = 12$  and  $S = 7$  almost matches the bulk liquid  
120 density at saturation, which is also plotted in Fig. 3 as reference. It is slightly  
121 smaller for  $S = 4.3$ . In the differential pressure the difference between the  
122 three simulations is more significant. The differential pressure in the center of  
123 the slab ( $y = 0$ ) almost reaches the zero line for  $S = 12$ , while for  $S = 7$  and  
124  $S = 4.3$  the pressure tensor is anisotropic throughout the liquid slab. The  
125 differential pressure can be seen as an indicator for the fluid to be isotropic  
126 or homogeneous, i.e. not influenced by the two interfaces. For  $S$  larger than  
127 12, the differential pressure fluctuates around zero in the center of the liquid  
128 slab for  $T = 0.7$ .

129 [Figure 4 about here.]

130 In Fig. 4, the plot of the differential pressure over the density is shown,  
131 which is obtained from the data displayed in Fig. 3. Additionally results  
132 for  $T = 1.0$  are shown. The plot in Fig. 4 exhibits van der Waals loops in  
133 all cases. The red squares and green stars correspond to a large liquid slab,

134 while the blue circles and brown triangles show the result of the smallest  
135 stable liquid slab. Like in Fig. 3, it is obvious that the differential pressure  
136 does not reach zero in the latter case.

137 The resulting surface tensions are shown in Tabs. 1 and 2. For all temper-  
138 atures the surface tension decreases when the liquid slab thickness decreases.

139 [Table 1 about here.]

140 [Table 2 about here.]

141 The differential pressure in the center of the liquid slab, i.e. the minimum  
142 differential pressure, is shown in Tabs. 3 and 4. As already discussed, the  
143 differential pressure in the center of the liquid slab increases with decreasing  
144 slab thickness. The density corresponding to the minimum differential pres-  
145 sure, i.e. the maximum density, in the liquid slab is shown in Tabs. 5 and 6.  
146 At the largest slab thickness the value agrees with the bulk properties. For  
147 lower slab thicknesses the density does not reach the bulk values. The mini-  
148 mum stable slab thickness increases with increasing temperature. Reducing  
149 the slab thickness below this point results in a rupture of the liquid phase  
150 and a transition from planar to cylindrical or spherical symmetry.

151 [Table 3 about here.]

152 [Table 4 about here.]

153 [Table 5 about here.]

154 [Table 6 about here.]



155

[Figure 5 about here.]

156 In Fig. 5, the relative surface tension – reduced by  $\gamma_\infty(T)$  as obtained  
 157 from the large slab simulations – is plotted over the slab thickness for differ-  
 158 ent temperatures. Confinement between two planar vapor-liquid interfaces  
 159 reduces the surface tension, and the numerical data suggest that this effect  
 160 is of the order  $1/S^3$ . In Fig. 6, the reduced density  $\rho'/\rho'_\infty$  is plotted over the  
 161 slab thickness for different temperatures. The relative density also decreases  
 162 upon decreasing the slab thickness and, similar to the surface tension, this  
 163 effect is approximately proportional to  $1/S^3$  and becomes more significant at  
 164 high temperatures.

165 For a slab thickness  $S > 12$ , the surface tension agrees with the value for  
 166 a large liquid slab, within the simulation uncertainty. For a fluid described  
 167 by the LJ potential, e.g. methane,<sup>59</sup> this means that confinement effects are  
 168 significant for slabs which are thinner than 4.5 nm. At high temperatures,  
 169 the density in the center of the liquid slab deviates more significantly from  
 170 the bulk value.

171

[Figure 6 about here.]

172

[Figure 7 about here.]

173 The present results suggest that the reduction of the density and the  
 174 surface tension due to confinement are related effects. In Fig. 7, the respec-  
 175 tive ratios are displayed together, which discloses an approximately linear  
 176 relationship. The regression

$$\frac{\rho'}{\rho'_\infty} \approx 0.76 \frac{\gamma}{\gamma_\infty} + 0.24 \quad (5)$$

177 is obtained from the simulation results.

#### 178 4. Conclusion

179 In the present work, molecular simulation was applied to study the influence  
180 of the slab thickness on the interfacial properties for planar vapor-liquid in-  
181 terfaces. The present results prove that such an effect exists for thin slabs  
182 and quantifies it for the LJ fluid. The surface tension decreases with de-  
183 creasing slab thickness, and so does the density in the middle of the slab.  
184 The differential pressure does not reach zero for liquid slabs smaller than 12,  
185 which proves that under such conditions, a bulk-like region is absent. The  
186 confinement effects for the surface tension and the density were found to scale  
187 with  $1/S^3$  in terms of the slab thickness  $S$ , so that a linear relation between  
188 both effects could be obtained.

189 The present results depart from those obtained by Weng et al.<sup>29</sup> in a  
190 previous study, where no systematic correlation between the slab thickness  
191 and the surface tension was found. For a LJ system at  $T = 0.818$ , Weng et  
192 al. detected minor fluctuations around a constant value ( $\gamma = 0.78 \pm 0.02$ ),  
193 without a clear tendency, for a range of slab thicknesses between  $S = 5.0$   
194 and  $9.0$ .<sup>29</sup> A juxtaposition with the present numerical data, cf. Tab. 1 and  
195 Fig. 5, according to which varying the slab thickness to such an extent has a  
196 significant influence on  $\gamma$ , clearly shows that there is a contradiction between  
197 present simulation results and the postulate of Weng et al. that “with film  
198 thickness ... surface tension values and density profiles show little varia-  
199 tion.”<sup>29</sup> For the simulations of Weng et al.,<sup>29</sup> however, no long-range cutoff  
200 correction was employed at all, and the computations were only carried out  
201 over 120 000 time steps, as opposed to a million time steps for the present  
202 series of simulations. Since systems with an interface relax more slowly than

203 the homogeneous bulk fluid, the extremely short simulation time could con-  
204 sstitute a serious limitation, affecting the accuracy of the results obtained by  
205 Weng et al.<sup>29</sup> to a significant extent.

206 The comparison with results from a recent study of Malijevský and Jack-  
207 son<sup>60</sup> suggests that the present results on confinement by two parallel planar  
208 vapor-liquid interfaces might also carry over qualitatively to confinement by  
209 the opposite sides of the single spherical interface that surrounds a small  
210 droplet. Therein, Malijevský and Jackson come to the conclusion that for  
211 liquid drops, the size dependence of the surface tension is best described  
212 by two distinct, additive terms: The conventional Tolman term, represent-  
213 ing curvature, which increases the surface tension (i.e. the Tolman length  
214 is found to be negative), as well as “an additional curvature dependence  
215 of the  $1/R^3$  form” which causes an eventual decrease of the surface tension  
216 “for smaller drops.”<sup>60</sup> Furthermore, Malijevský and Jackson observe that the  
217 characteristic droplet radius, below which this negative corrective term be-  
218 comes dominant, “increases with increasing  $r_c$ ” and conjecture that “such a  
219 crossover occurs when ... no ‘bulk’ region can be assigned inside the drop.  
220 In this case even particles in the centre of the drop ‘feel’ the interface.”

221 The present results lend further plausibility to this conjecture of Mal-  
222 ijevský and Jackson. There could be a relation between their  $1/R^3$  term  
223 and the  $1/S^3$  confinement effect from the present study. According to such  
224 a hypothesis, these contributions would both represent the deviation from  
225 bulk-like behavior of the liquid phase due to confinement.

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234 largest slab were carried out on the XC 4000 supercomputer at the Steinbuch  
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## 236 **References**

- 237 1. Yaguchi H., Yano T., Fujikawa S. *J Fluid Sci Technol* 2010;**5**(2):180–  
238 91.
- 239 2. Thompson S.M., Gubbins K.E., Walton J.P.R.B., Chantry R.A.R.,  
240 Rowlinson J.S. *J Chem Phys* 1984;**81**(1):530–42.
- 241 3. Blokhuis E.M., Bedeaux D. *J Chem Phys* 1992;**97**(5):3576–86.
- 242 4. Onischuk A.A., Vosel S.V., Borovkova O.V., Baklanov A.M., Karasev  
243 V.V., di Stasio S. *J Chem Phys* 2012;**136**:224506.
- 244 5. Tolman R.C. *J Chem Phys* 1949;**17**(3):333–7.
- 245 6. Haye M.J., Bruin C. *J Chem Phys* 1994;**100**(1):556–9.
- 246 7. Horsch M., Hasse H., Shchekin A.K., Agarwal A., Eckelsbach S.,  
247 Vrabec J., Müller E.A., Jackson G. *Phys Rev E* 2012;**85**:031605.

- 248 8. Zhu R.Z., Yan H. *Chin Phys B* 2011;**20**:016801.
- 249 9. Tröster A., Oettel M., Block B., Virnau P., Binder K. *J Chem Phys*  
250 2012;**136**:064709.
- 251 10. Buff F.P. *J Chem Phys* 1951;**19**(12):1591–4.
- 252 11. Lovett R. *Rep Prog Phys* 2007;**70**(2):195–218.
- 253 12. Vrabec J., Horsch M., Hasse H. *J Heat Transfer (ASME)* 2009;  
254 **131**:043202.
- 255 13. Kraska T. *J Chem Phys* 2006;**124**:054507.
- 256 14. Chkonia G., Wölk J., Strey R., Wedekind J., Reguera D. *J Chem*  
257 *Phys* 2009;**130**:064505.
- 258 15. Baidakov V.G., Kaverin A.M., Andbaeva V.N. *J Phys Chem B* 2008;  
259 **112**(41):12973–5.
- 260 16. Bolhuis P.G., Chandler D. *J Chem Phys* 2000;**113**(18):8154–60.
- 261 17. Schmelzer J.W.P., Boltachev G.S., Baidakov V.G. *J Chem Phys* 2006;  
262 **124**:194503.
- 263 18. van Erp T.S., Moroni D., Bolhuis P.G. *J Chem Phys* 2003;  
264 **118**(17):7762–74.
- 265 19. McGinty D.J. *J Chem Phys* 1973;**58**(11):4733–42.
- 266 20. Lee J.K., Barker J.A., Abraham F.F. *J Chem Phys* 1973;**58**(8):3166–  
267 80.

- 268 21. Orea P., López Lemus J., Alejandre J. *J Chem Phys* 2005;**123**:114702.
- 269 22. Gloor G.J., Jackson G., Blas F.J., de Miguel E. *J Chem Phys* 2005;  
270 **123**:134703.
- 271 23. Janeček J. *J Chem Phys* 2009;**131**:124513.
- 272 24. Werner A., Schmid F., Müller M., Binder K. *J Chem Phys* 1997;  
273 **107**:8175–88.
- 274 25. Feder J., Russell K.C., Lothe J., Pound G.M. *Adv Phys* 1966;  
275 **15**(1):111–78.
- 276 26. Lekkerkerker H.N.W., de Villeneuve V.W.A., de Folter J.W.J.,  
277 Schmidt M., Hennequin Y., Bonn D., Indekeu J.O., Aarts D.G.A.L.  
278 *Eur Phys J B* 2008;**64**(3-4):341–7.
- 279 27. Sengers J.V., van Leeuwen J.M.J. *Phys Rev A* 1989;**39**(12):6346–55.
- 280 28. Gelfand M.P., Fisher M.E. *Physica A* 1990;**166**(1):1–74.
- 281 29. Weng J.G., Park S., Lukes J., Tien C.L. *J Chem Phys* 2000;  
282 **113**(14):5917–23.
- 283 30. Vrabec J., Hasse H. *Molec Phys* 2002;**100**(21):3375–83.
- 284 31. Möller D., Fischer J. *Molec Phys* 1990;**69**(3):463–73.
- 285 32. Panagiotopoulos A.Z. *Molec Phys* 1987;**61**(4):813–26.
- 286 33. Salomons E., Mareschal M. *J Phys: Condens Matter* 1991;**3**(20):3645–  
287 61.

- 288 34. Ghoufi A., Malfreyt P. *J Chem Phys* 2012;**136**:024104.
- 289 35. Watanabe H., Ito N., Hu C.K. *J Chem Phys* 2012;**136**:204102.
- 290 36. Hamada Y., Koga K., Tanaka H. *Physica A* 2009;**388**:2289–98.
- 291 37. Binder K. *Physica A* 2003;**319**:99–114.
- 292 38. Lion T.W., Allen R.J. *J Phys: Condens Matter* 2012;**24**:284133.
- 293 39. Mareschal M., Baus M., Lovett R. *J Chem Phys* 1997;**106**(2):645–54.
- 294 40. Imre A.R., Kraska T. *Fluid Phase Equilib* 2009;**284**(1):31–7.
- 295 41. Kraska T., Römer F., Imre A. *J Phys Chem B* 2009;**113**(14):4688–97.
- 296 42. Alejandre J., Chapela G.A. *J Chem Phys* 2010;**132**:014701.
- 297 43. Mansfield K.F., Theodorou D.M. *Macromolecules* 1991;**24**(15):4295–  
298 309.
- 299 44. de Gregorio R., Benet J., Katcho N.A., Blas F.J., MacDowell L.G.  
300 *J Chem Phys* 2012;**136**:104703.
- 301 45. van Giessen A.E., Blokhuis E.M. *J Chem Phys* 2009;**131**:164705.
- 302 46. Vrabec J., Kedia G.K., Fuchs G., Hasse H. *Molec Phys* 2006;  
303 **104**(9):1509–27.
- 304 47. Ge J., Kjelstrup S., Bedeaux D., Simon J.M., Rousseau B. *Phys Rev*  
305 *E* 2007;**75**:061604.
- 306 48. Neumann M. *J Chem Phys* 1985;**82**(12):5663–72.

- 307 49. Onsager L. *J Am Chem Soc* 1936;**58**(8):1486–93.
- 308 50. Janeček J. *J Phys Chem B* 2006;**110**(12):6264–9.
- 309 51. Lotfi A., Vrabec J., Fischer J. *Molec Sim* 1990;**5**(3-4):233–43.
- 310 52. Mecke M., Winkelmann J., Fischer J. *J Chem Phys* 1997;  
311 **107**(21):9264–70.
- 312 53. Shen V.K., Mountain R.D., Errington J.R. *J Phys Chem B* 2007;  
313 **111**(22):6198–207.
- 314 54. Walton J., Tildesley D., Rowlinson J., Henderson J. *Molec Phys*  
315 1983;**48**(6):1357–68.
- 316 55. Irving J., Kirkwood J. *J Chem Phys* 1949;**17**(3):338–43.
- 317 56. Buchholz M., Bungartz H.J., Vrabec J. *J Computational Sci* 2011;  
318 **2**(2):124–9.
- 319 57. Fincham D. *Molec Phys* 1992;**8**(3-5):165–78.
- 320 58. Pérez Pellitero J., Ungerer P., Orkoulas G., Mackie A.D. *J Chem*  
321 *Phys* 2006;**125**:054515.
- 322 59. Vrabec J., Stoll J., Hasse H. *J Phys Chem B* 2001;**105**:12126–33.
- 323 60. Malijevský A., Jackson G. *J Phys: Cond Mat* 2012;**24**:464121.
- 324 61. Holcomb C.D., Clancy P., Zollweg J.A. *Molec Phys* 1993;**78**(2):437–59.
- 325 62. Potoff J.J., Panagiotopoulos A.Z. *J Chem Phys* 2000;**112**(14):6411–5.



326 63. in 't Veld P.J., Ismail A.E., Grest G.S. *J Chem Phys* 2007;**127**:144711.

327 64. López Lemus J., Alexandre J. *Molec Phys* 2002;**100**(18):2983–92.

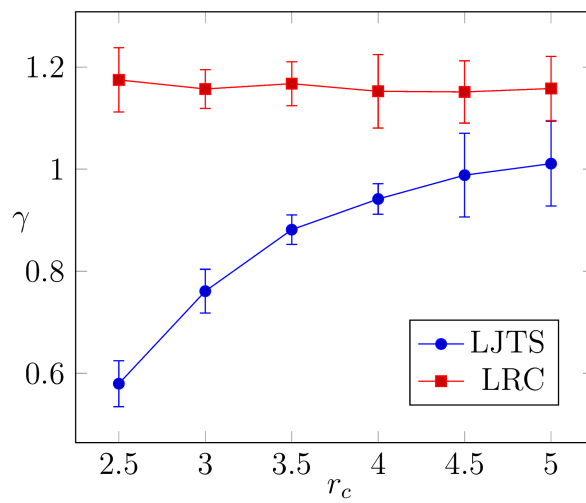


Figure 1: Dependence of the computed surface tension  $\gamma$  on the employed cutoff radius  $r_c$ , for the truncated and shifted LJ potential without long-range effects (LJTS) and for the full LJ potential with a long-range correction (LRC) according to Janeček.<sup>50</sup> The simulations were carried out in the canonical ensemble with  $N = 2\,048$  particles at  $T = 0.7$  using a simulation box with an elongation of  $\ell_y = 66$  in the direction perpendicular to the vapor-liquid interfaces.

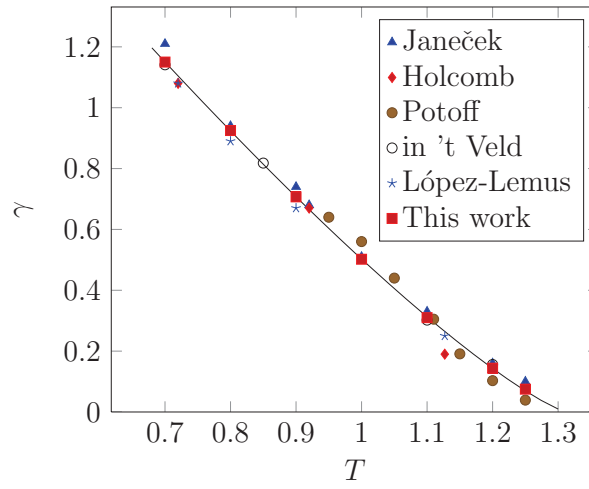


Figure 2: Surface tension  $\gamma$  over the temperature for large liquid slabs. Comparison of results of different authors: (red squares) This work, (blue triangles) Janeček,<sup>50</sup> (red diamonds) Holcomb et al.,<sup>61</sup> (brown circles) Potoff and Panagiotopoulos,<sup>62</sup> (black circles) in't Veld et al.,<sup>63</sup> (blue stars), as well as López Lemus and Alejandre,<sup>64</sup>; (solid line) regression from Eq. (4).

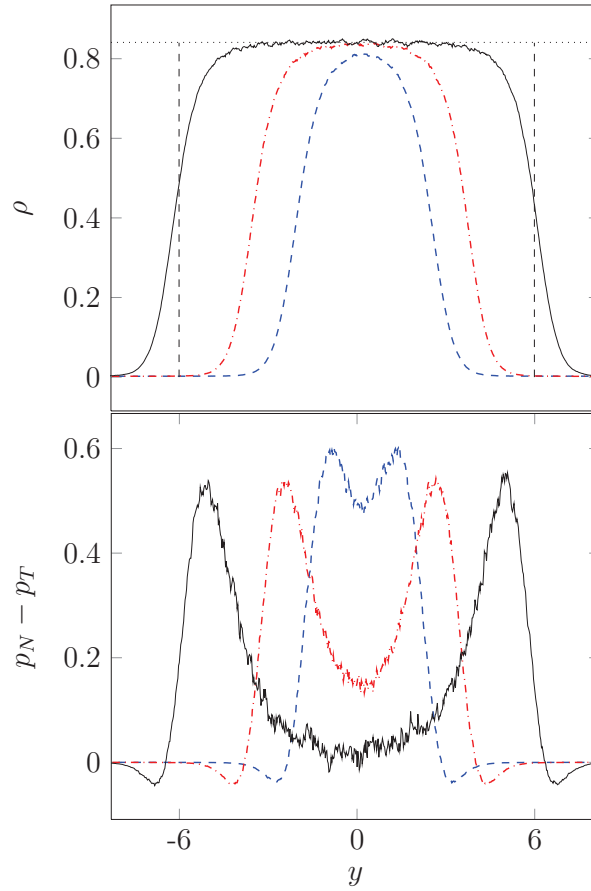


Figure 3: Density  $\rho$  (top) and differential pressure  $p_N - p_T$  (bottom) over the  $y$  coordinate (i.e. the direction perpendicular to the interface). The temperature is  $T = 0.7$ . The blue dashed line corresponds to the minimum stable configuration which is  $S = 4.3$  for this temperature, while the red dash dotted one corresponds to  $S = 7$  and the black solid one to  $S = 12$ . The dotted line in the upper picture represents the bulk liquid density and the difference between the vertical dashed lines in the upper picture represent the equimolar slab thickness.

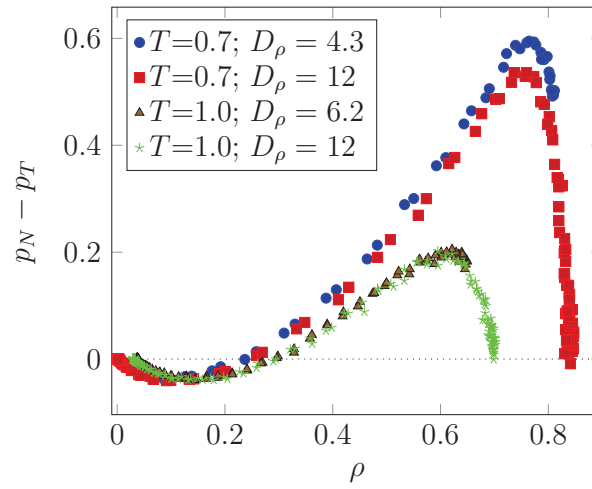


Figure 4: Differential pressure  $p_N - p_T$  over the density for the temperature  $T = 0.7$  and  $T = 1.0$ . The blue circles and brown triangles correspond to the minimum stable configuration for the corresponding temperature, while the red squares and green stars correspond to  $S = 12$ . The dotted line represents the zero line.

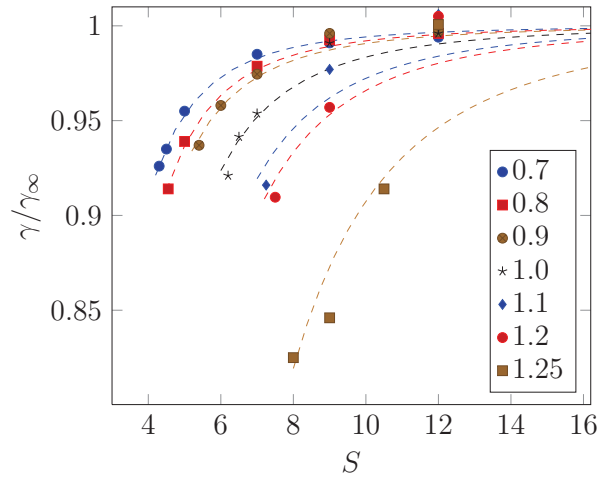


Figure 5: Reduced surface tension  $\gamma/\gamma_\infty$  over the slab thickness  $S$  for different temperatures. The dashed lines represent the expression  $\gamma/\gamma_\infty = 1 - a(T)/S^3$ , where the temperature-dependent coefficients were adjusted to the simulation results, yielding  $a(0.7) = 5.8$ ,  $a(0.8) = 7.9$ ,  $a(0.9) = 9.3$ ,  $a(1.0) = 16$ ,  $a(1.1) = 28$ ,  $a(1.2) = 34$ , and  $a(1.25) = 93$ .

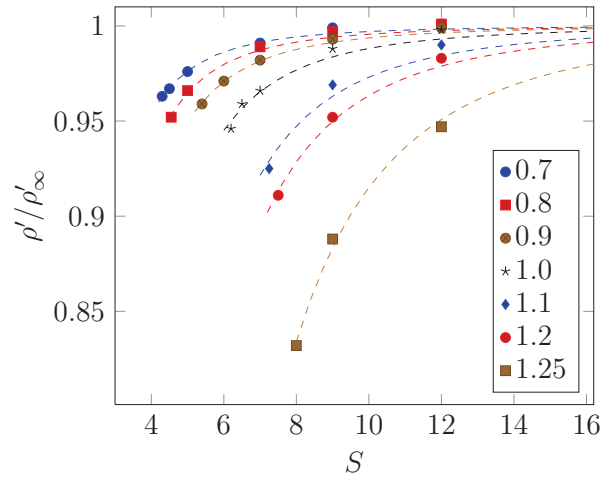


Figure 6: Reduced density  $\rho'/\rho'_\infty$  over the slab thickness  $S$  for different temperatures. The dashed lines represent the expression  $\rho'/\rho'_\infty = 1 - b(T)/S^3$ , where the temperature-dependent coefficients were adjusted to the simulation results, yielding  $b(0.7) = 3.0$ ,  $b(0.8) = 4.4$ ,  $b(0.9) = 6.3$ ,  $b(1.0) = 12$ ,  $b(1.1) = 27$ ,  $b(1.2) = 37$ , and  $b(1.25) = 85$ .

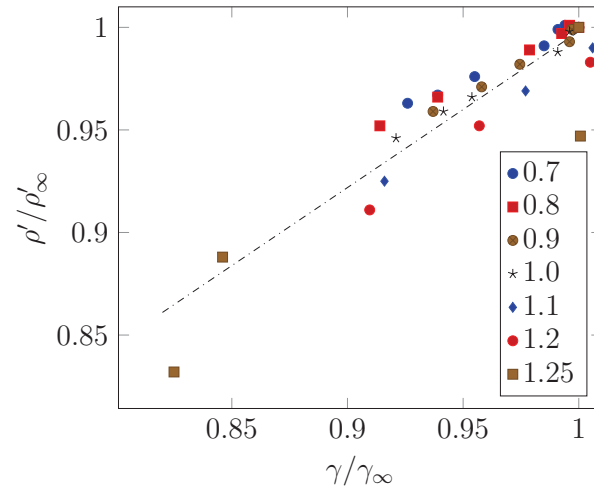


Figure 7: Reduced density  $\rho'/\rho'_\infty$  over the reduced surface tension  $\gamma/\gamma_\infty$  for different temperatures. The dash dotted line represents the regression from Eq. (5).



$T$	$N$	$\ell_y$	$\ell_{\parallel}$	$S$	$\gamma$
0.7	300 000	100	94.2	40.0	1.150(4)
	16 000	50.0	39.7	12.0	1.14(7)
	16 000	50.0	45.8	9.0	1.14(6)
	16 000	50.0	51.8	7.0	1.13(3)
	16 000	50.0	61.1	5.0	1.10(2)
	16 000	50.0	64.3	4.5	1.08(1)
	16 000	50.0	65.7	4.3	1.06(2)
0.8	300 000	100	96.4	40.0	0.93(1)
	16 000	50.0	40.3	12.0	0.92(6)
	16 000	50.0	46.4	9.0	0.92(3)
	16 000	50.0	52.3	7.0	0.91(3)
	16 000	50.0	61.2	5.0	0.87(3)
	16 000	50.0	63.9	4.55	0.85(2)
0.9	300 000	100	98.2	40.0	0.707(8)
	16 000	50.0	40.8	12.0	0.71(5)
	16 000	50.0	46.5	9.0	0.70(4)
	16 000	50.0	51.9	7.0	0.69(3)
	16 000	50.0	55.4	6.0	0.68(2)
	16 000	50.0	58.1	5.4	0.66(3)

Table 1: Surface tension  $\gamma$  in dependence of the slab thickness  $S$  (low temperatures). The total elongation of the simulation box is indicated as  $\ell_y$  in the direction perpendicular to the vapor-liquid interfaces and as  $\ell_{\parallel}$  ( $= \ell_x = \ell_z$ ) in the other spatial directions. The statistical error in terms of the final digit is shown in parentheses.

$T$	$N$	$\ell_y$	$\ell_{\parallel}$	$S$	$\gamma$
1.0	300 000	100	100	40.0	0.502(5)
	16 000	50.0	40.1	12.0	0.50(6)
	16 000	50.0	46.1	9.0	0.50(2)
	16 000	50.0	50.8	7.0	0.48(5)
	16 000	50.0	53.0	6.2	0.5(1)
1.1	300 000	100	102	40.0	0.310(4)
	16 000	50.0	40.9	12.0	0.31(4)
	16 000	50.0	45.3	9.0	0.30(3)
	16 000	50.0	48.6	7.25	0.28(4)
	16 000	50.0	50.2	6.5	0.26(3)
1.2	300 000	100	101	40.0	0.144(8)
	16 000	50.0	38.2	12.0	0.14(5)
	16 000	50.0	40.8	9.0	0.14(4)
	16 000	50.0	42.4	7.5	0.13(4)
1.25	300 000	100	101	40.0	0.075(4)
	16 000	50.0	37.2	12.0	0.08(5)
	16 000	50.0	39.2	9.0	0.06(2)
	16 000	50.0	39.9	8.0	0.06(4)

Table 2: Surface tension  $\gamma$  in dependence of the slab thickness  $S$  (high temperatures), cf. Tab. 1.

$S$	$T$			
	0.7	0.8	0.9	1.0
12	-0.01 (6)	-0.01 (4)	0.00 (3)	0.00 (2)
9	0.04 (6)	0.04 (4)	0.04 (2)	0.04 (5)
7	0.13 (5)	0.12 (3)	0.12 (4)	0.11 (5)
6.2	n/a	n/a	n/a	0.16 (5)
6	n/a	n/a	0.18 (2)	*
5.4	n/a	n/a	0.24 (3)	*
5	0.34 (2)	0.32 (3)	*	*
4.55	n/a	0.38 (6)	*	*
4.5	0.43 (4)	*	*	*
4.3	0.48 (3)	*	*	*

Table 3: Differential pressure  $p_N - p_T$  in the center of the liquid slab in dependence on the slab thickness  $S$  (low temperatures). “n/a”: data not determined; asterisks: liquid slab found to be unstable.

$S$	$T$		
	1.1	1.2	1.25
12	0.01 (2)	0.00 (5)	0.00 (3)
9	0.05 (2)	0.02 (4)	0.01 (3)
8	n/a	n/a	0.02 (2)
7.5	n/a	0.03 (3)	*
7.25	0.10 (3)	*	*
6.5	0.12 (2)	*	*

Table 4: Differential pressure  $p_N - p_T$  in the center of the liquid slab (high temperatures), cf. Tab. 3.

$S$	$T$			
	0.7	0.8	0.9	1.0
40	0.8410 (2)	0.7974 (3)	0.7507 (6)	0.699 (1)
12	0.84 (3)	0.80 (2)	0.75 (1)	0.70 (1)
9	0.84 (1)	0.80 (1)	0.75 (1)	0.69 (2)
7	0.83 (1)	0.79 (1)	0.74 (2)	0.68 (2)
6.2	n/a	n/a	n/a	0.66 (5)
6	n/a	n/a	0.73 (2)	*
5.4	n/a	n/a	0.72 (1)	*
5	0.82 (1)	0.77 (1)	*	*
4.55	n/a	0.76 (3)	*	*
4.5	0.81 (1)	*	*	*
4.3	0.81 (1)	*	*	*

Table 5: Density  $\rho'$  in the center of the liquid slab in dependence on the slab thickness  $S$  (low temperatures).

$S$	$T$		
	1.1	1.2	1.25
40	0.6393 (4)	0.564 (2)	0.515 (7)
12	0.63 (3)	0.56 (3)	0.49 (10)
9	0.62 (2)	0.54 (3)	0.46 (4)
8	n/a	n/a	0.43 (15)
7.5	n/a	0.51 (7)	*
7.25	0.59 (3)	*	*
6.5	0.57 (8)	*	*

Table 6: Density  $\rho'$  in the center of the liquid slab (high temperatures), cf. Tab. 5.