

BCNucleation-Aggregation Workshop

Grand canonical molecular dynamics simulation of homogeneous nucleation

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Molecular simulation of nucleation

Indirect simulation:

Transition path sampling Determination of the critical size ... by observing single droplets in non-equilibrium ... by observing single droplets in equilibrium

Direct simulation:

... of a metastable state far from the spinodal line ... of nucleation at a high supersaturation, decreasing over time

... of a metastable state near the spinodal line ... of nucleation at a constantly high supersaturation





The critical nucleus

... is defined by a *stable* or *unstable* equilibrium with the vapor.



It is essential to know the supersaturation in terms of $\Delta \mu$.





Equilibrium vapor pressure



Equilibrium condition for a droplet containing *j* molecules:

p = p(T, j)

 ΔG at constant *p* and *T*: 1 unstable equilibrium

 ΔF at constant V and T:

1 unstable equilibrium
1 stable equilibrium





Systems containing a single droplet

- Vapor and liquid are equilibrated separately.
- A small (j < 10000) droplet is inserted into the vapor.
- If the droplet cannot evaporate completely, an equilibrium is established within a few nanoseconds.



t-s-LJ fluid (r_c = 2.5 σ)





Surface tension

Integration of the $p_N(r)$ profile:

$$\gamma^{-3} = \frac{-(p_{\rm I}-p)^2}{8} \int_0^\infty r^3 \left[\frac{dp_{\rm N}(r)}{dr}\right] dr$$

Size dependence (Tolman):

$$\frac{\gamma_{\infty}}{\gamma} = 1 + \frac{2\delta_{\mathrm{T}}}{R_{\gamma}} + O(R_{\gamma}^{-2})$$

Correlation from simulation data for $T = 0.65, 0.70, \dots 0.95 \epsilon/k$:

$$\frac{\delta_{\rm T}}{R_{\rm Y}} = \left(\frac{0.7}{1 - T/T_{\rm c}} - 0.9\right) j^{-1/3}$$





Tolman equation



The higher order terms of the Tolman equation should not be neglected.





Direct MD simulation of nucleation

- Integration time step typically between 1 and 5 fs; Feasible simulation time: on the order of nanoseconds.
- > A saturated vapor with $V = 10^{-20}$ m³ contains: 800,000 molecules (methane at 114 K = 0.6 T_c) 7,000,000 molecules (CO₂ at 253 K = 0.83 T_c)
- > Minimal nucleation rate accessible by direct simulation:







Grand canonical molecular dynamics

Algorithm according to Cielinski:

- fixed values of μ , V und T
- test insertion of a molecule at a random position

$$\boldsymbol{P}_{\text{ins}} = \max\left[1, \exp\left(\frac{\mu - \Delta U_{\text{ins}}}{kT}\right) \frac{V}{\Lambda^3 (N+1)}\right]$$

• test deletion of a random molecule

$$P_{\rm del} = \max\left[1, \exp\left(\frac{-\mu - \Delta U_{\rm ins}}{kT}\right)\frac{V}{\Lambda^3 N}\right]$$

• equal number of test insertions and deletions $(10^{-5} - 10^{-3} / \text{step})$





Supersaturation from *NVT* and μVT simulation







Szilárd's demon





SZILÁRD





McDonald's demon







Interactive presentation: McDonald's demon







Comparison: *NVT* and μVT simulation







Nucleus size distribution

t-s-LJ fluid at $T = 0.7 \epsilon/k$: $\mu VT (S = 2.866)$ and $NVT (\rho = 0.004044 \sigma^3)$ simulation



Good agreement with CNT for *j*^{*} and the number of small nuclei.





Threshold dependence of the intervention rate



CNT predicts an acceptable value for j^* and underestimates J significantly.





GCMD simulation of nucleation: Results







Conclusion

- MD simulation of **equilibria** allows sampling over an arbitrary time interval, eventually leading to the desired level of accuracy.
- Single droplets can be stable in the canonical ensemble.
- A supersaturated vapor near the spinodal line can be stabilized by grand canonical simulation with McDonald's demon.
- The **classical theory** leads to acceptable results for the t-s-LJ fluid. However, it does not take into account curvature effects on the surface tension.

